

# Literature Report VI

## Organocatalytic Dynamic Kinetic Resolution Enabled Asymmetric Synthesis of Phosphorus-Containing Chiral Helicenes

---

Reporter: Jian Chen

Checker: Qing-Xian Xie

Date: 2023-12-25

Wu, J.-Q.; Fang, S.; Zheng, X.; He, J.; Ma, Y.; Su, Z.; [Wang, T.](#)

*Angew. Chem. Int. Ed.* **2023**, 62, e202309515

# CV of Prof. Wang Tianli (王天利)

---



## Research:

- ❑ Catalytic synthesis methodology
- ❑ Green chemistry
- ❑ Design and application of phosphonium salt catalyst

---

## Background:

- ❑ **2002-2006** B.S., Sichuan University (Prof. Feng Xiaoming)
- ❑ **2006-2011** Ph.D., ICCAS (Prof. Fan Qing-Hua)
- ❑ **2011-2012** Assistant Professor, ICCAS
- ❑ **2012-2016** Postdoc., National University of Singapore (Prof. Lu Yixin)
- ❑ **2016-Now** Professor, Sichuan University

# Contents

---

1

**Introduction**

---

2

**Organocatalytic Asymmetric Synthesis of Phosphorus-Containing Chiral Helicenes**

---

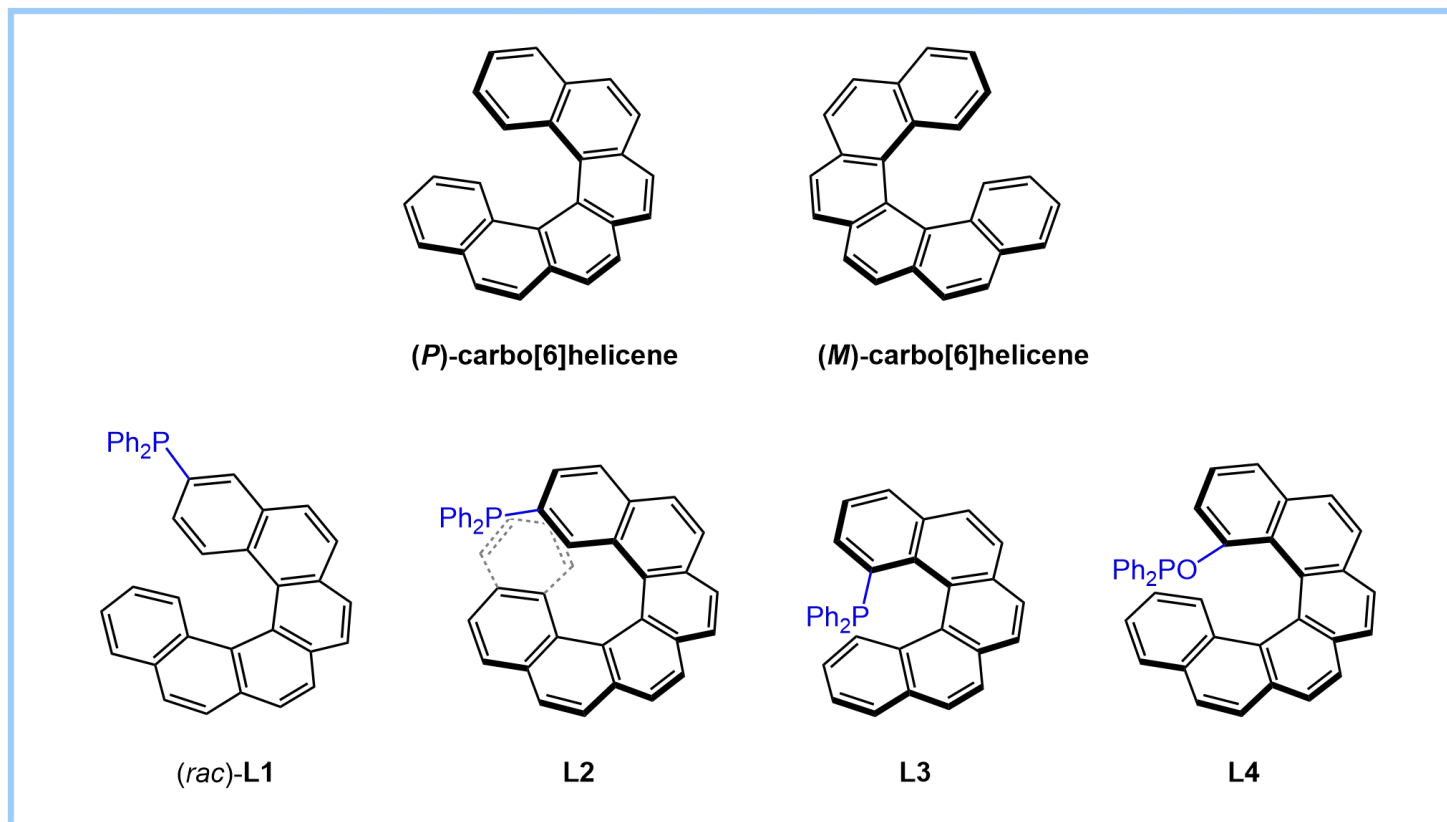
3

**Summary**

---

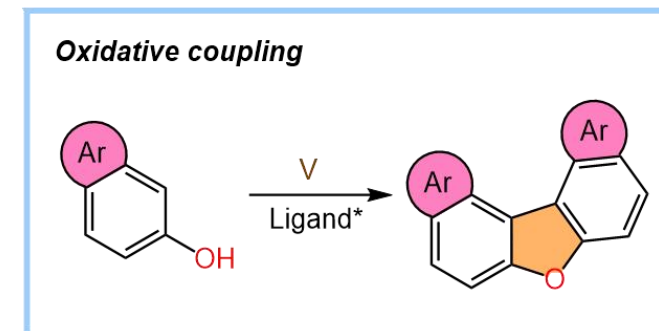
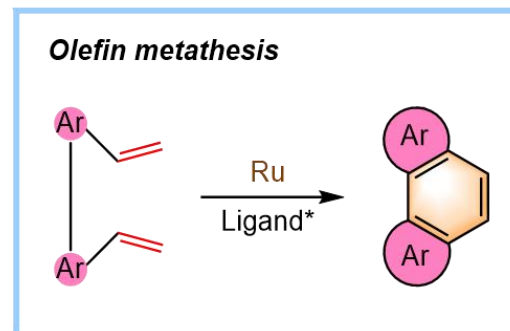
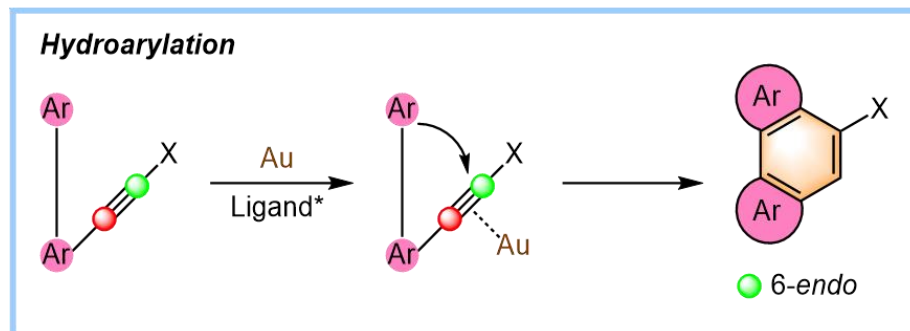
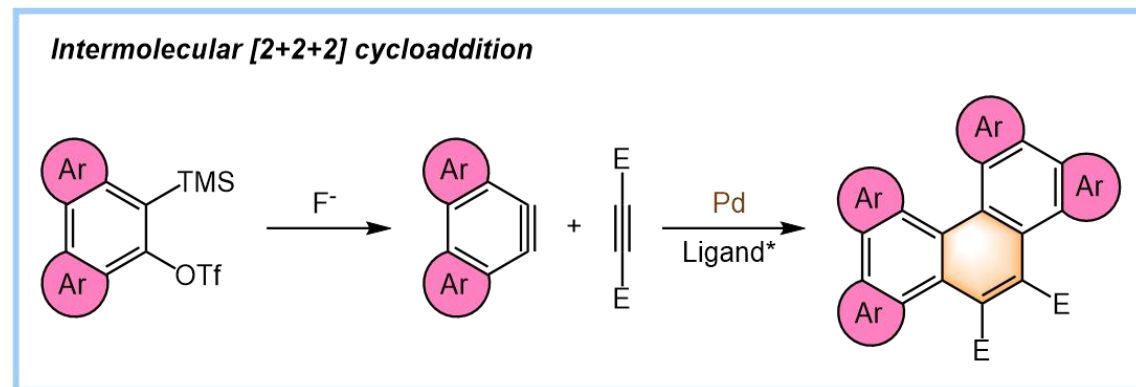
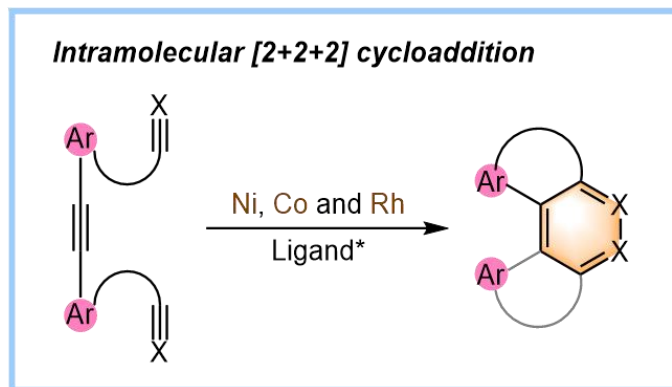
# Introduction

## Representative monophosphine-containing helicenes



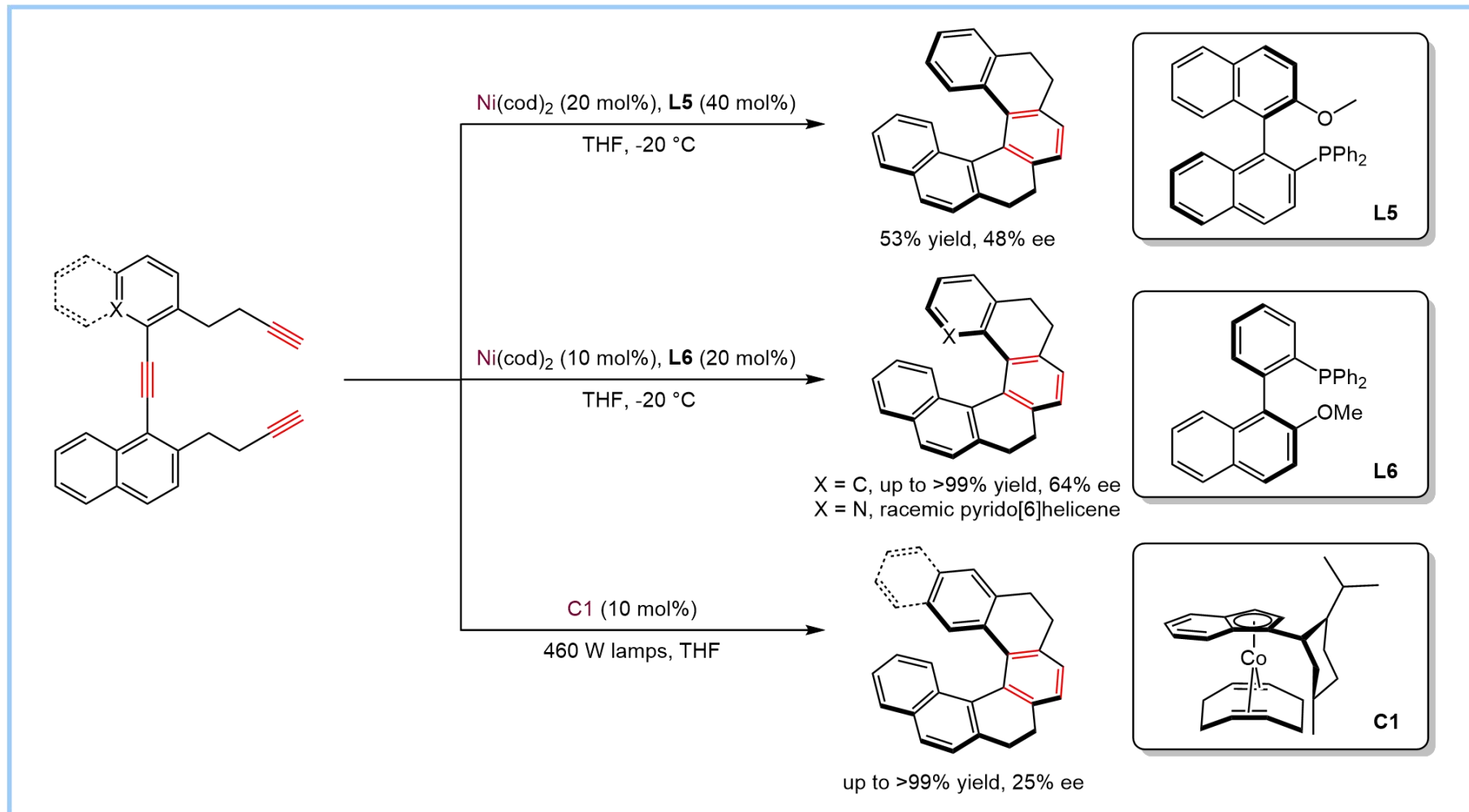
Wang, Y.; Wu, Z.-G.; Shi, F.\* *Chem Catalysis* **2022**, 2, 3077-3111

# Metal-Catalyzed Synthesis of Helicenes



# Metal-Catalyzed Synthesis of Helicenes

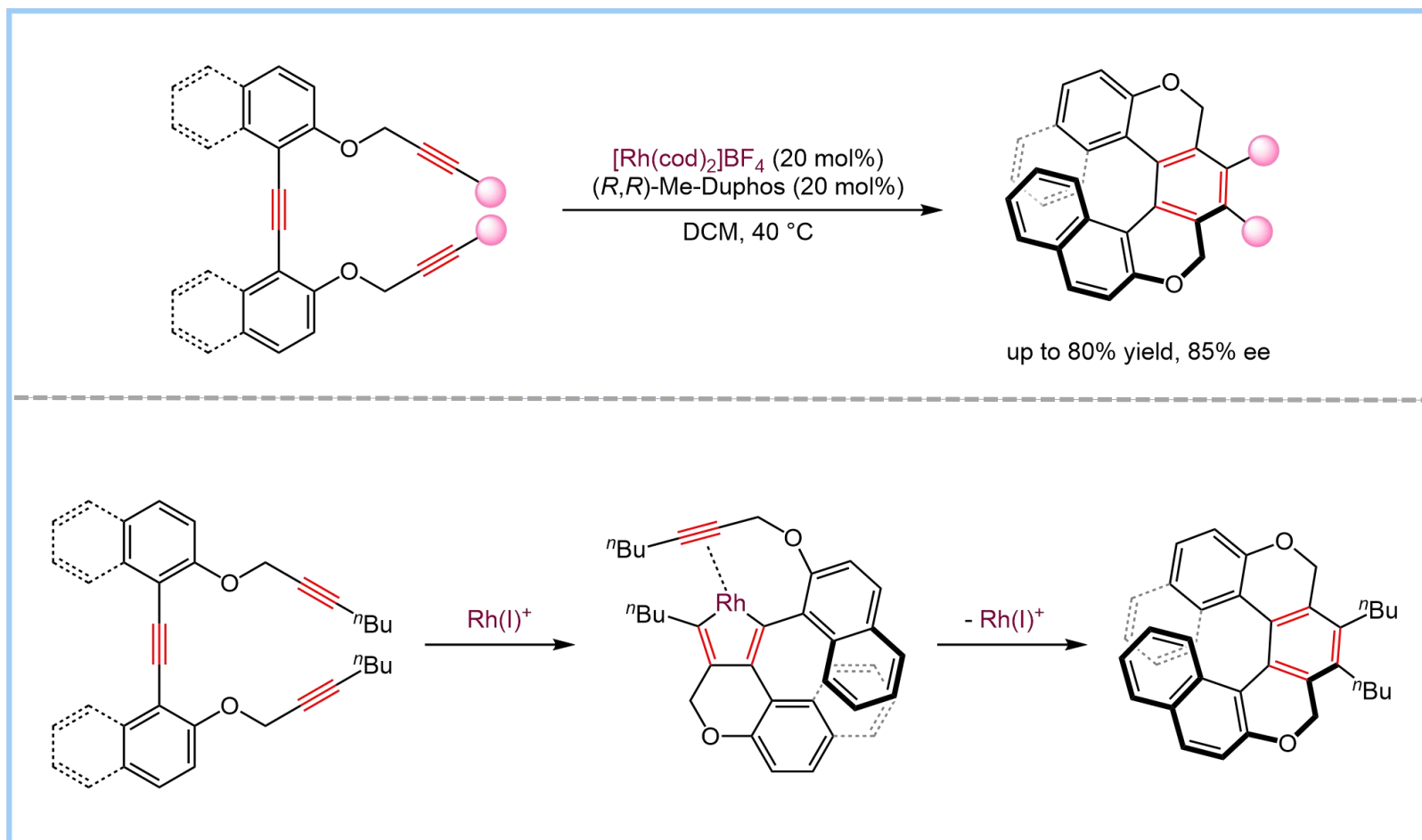
## Intramolecular [2+2+2] cycloaddition (**Co** and **Ni**)



Stara, I. G. *et al.* *Tetrahedron Letters* **1999**, 40, 1993-1996; Heller, B. *et al.* *Journal of Organometallic Chemistry* **2013**, 723, 98-102

# Metal-Catalyzed Synthesis of Helicenes

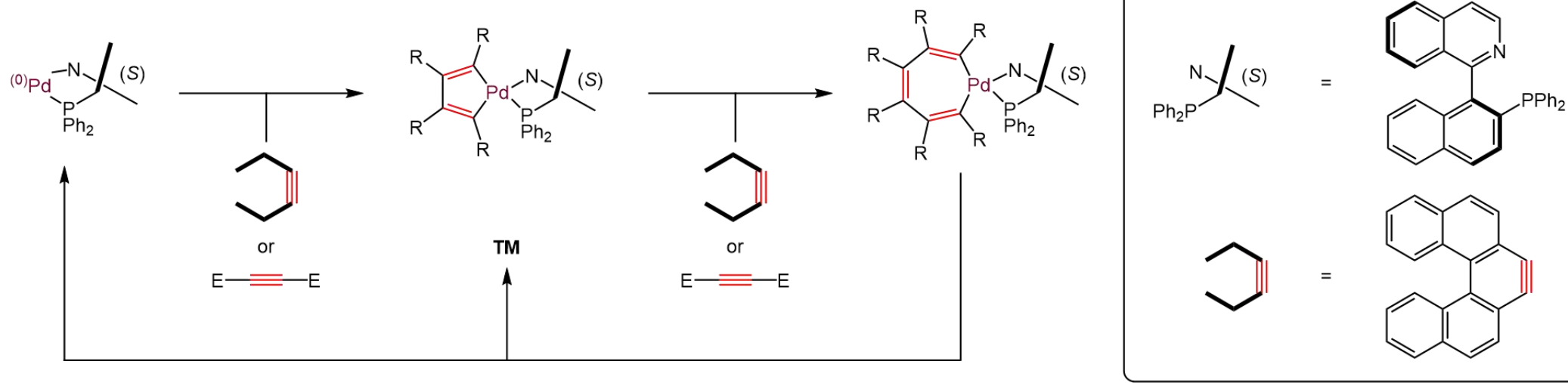
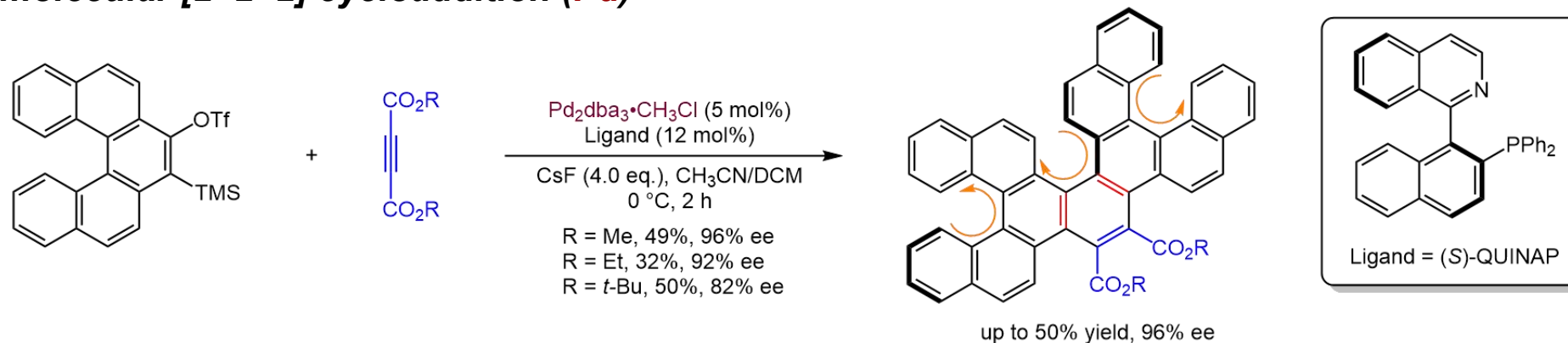
## Intramolecular [2+2+2] cycloaddition (*Rh*)



Tanaka, K.\*; Kamisawa, A.; Suda, T.; Noguchi, K.; Hirano, M. *J. Am. Chem. Soc.* **2007**, *129*, 12078-12079

# Metal-Catalyzed Synthesis of Helicenes

## Intermolecular [2+2+2] cycloaddition (Pd)

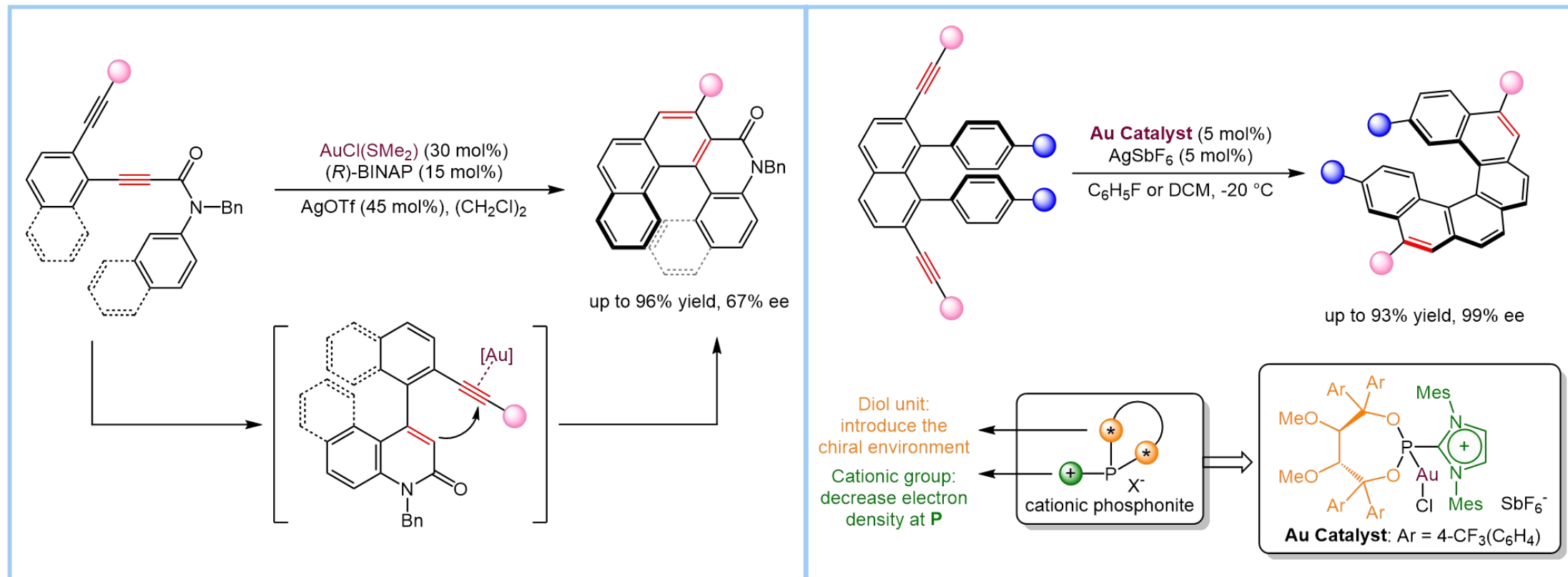


Yubuta, A.; Hosokawa, T.; Gon, M.; Tanaka, K.; Chujo, Y.; Tsurusaki, A.; Kamikawa, K.\* *J. Am. Chem. Soc.* **2020**, *142*, 10025-10033



# Metal-Catalyzed Synthesis of Helicenes

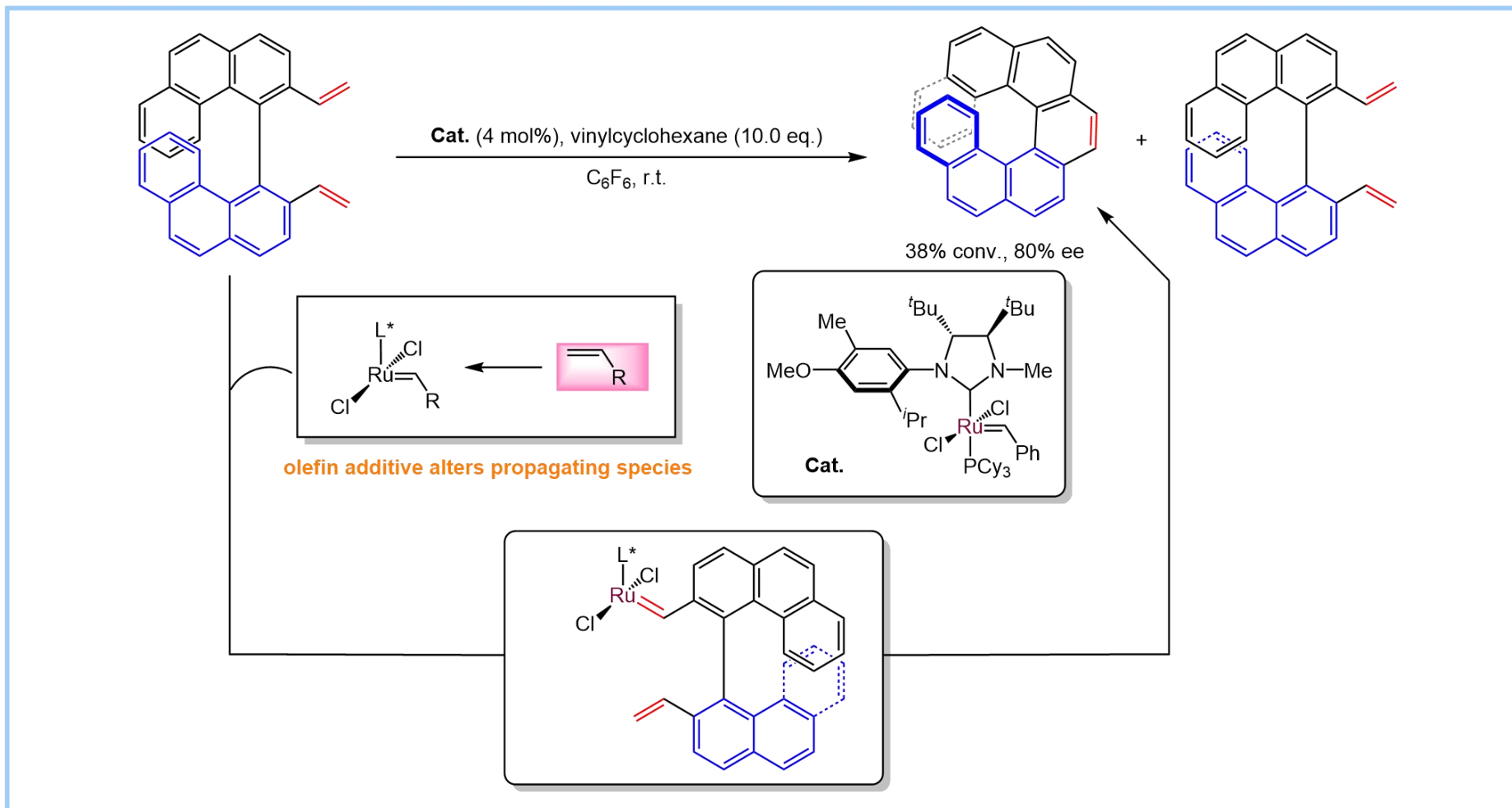
## Hydroarylation (Au)



Nakamura, K.; Furumi, S.; Takeuchi, M.; Shibuya, T.; Tanaka, K.\* *J. Am. Chem. Soc.* **2014**, *136*, 5555-5558  
Gonzalez-Fernandez, E.; Fares, C.; Lehmann, C. W.; Alcarazo, M.\* *J. Am. Chem. Soc.* **2017**, *139*, 1428-1431

# Metal-Catalyzed Synthesis of Helicenes

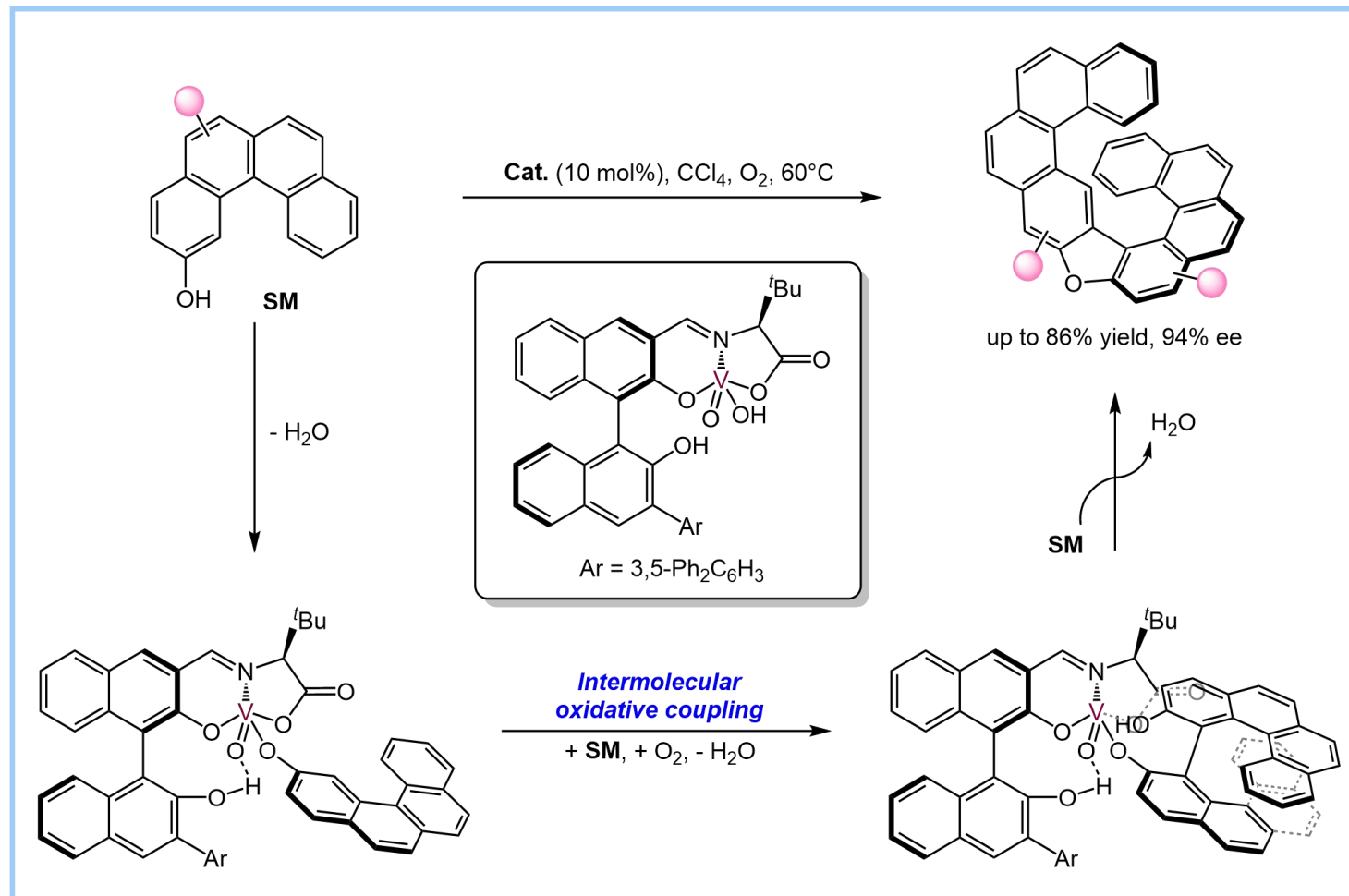
## Olefin metathesis (*Ru*)



Grandbois, A.; Collins, S. K.\* *Chem. Eur. J.* **2008**, *14*, 9323-9329

# Metal-Catalyzed Synthesis of Helicenes

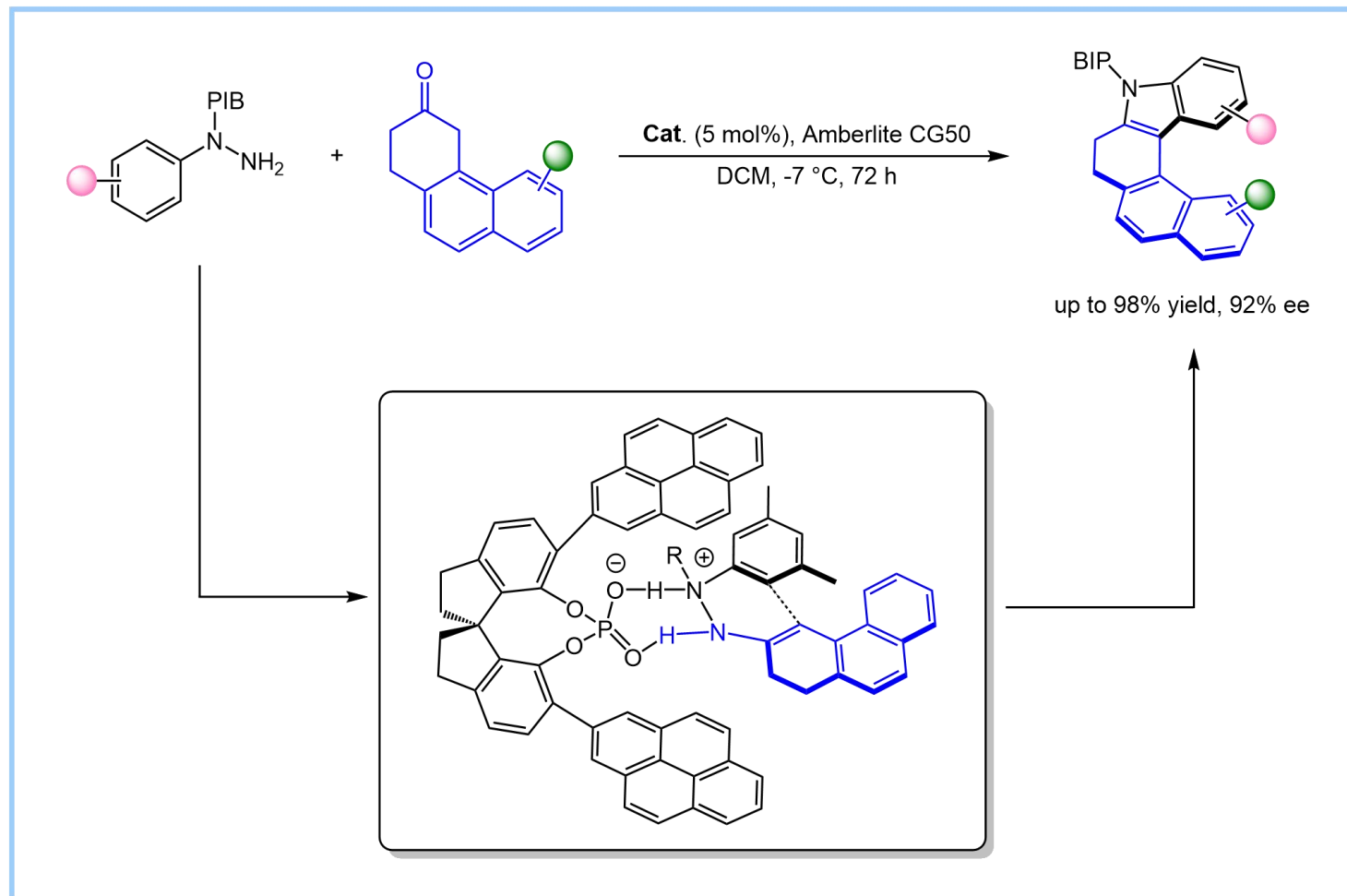
## Oxidative coupling (V)



Sako, M.; Takeuchi, Y.; Tsujihara, T.; Kodera, J.; Chujo, Y.; Kawano, T.; Sasai, H.\* *J. Am. Chem. Soc.* **2016**, *138*, 11481-11484

# Organocatalytic Synthesis of Helicenes

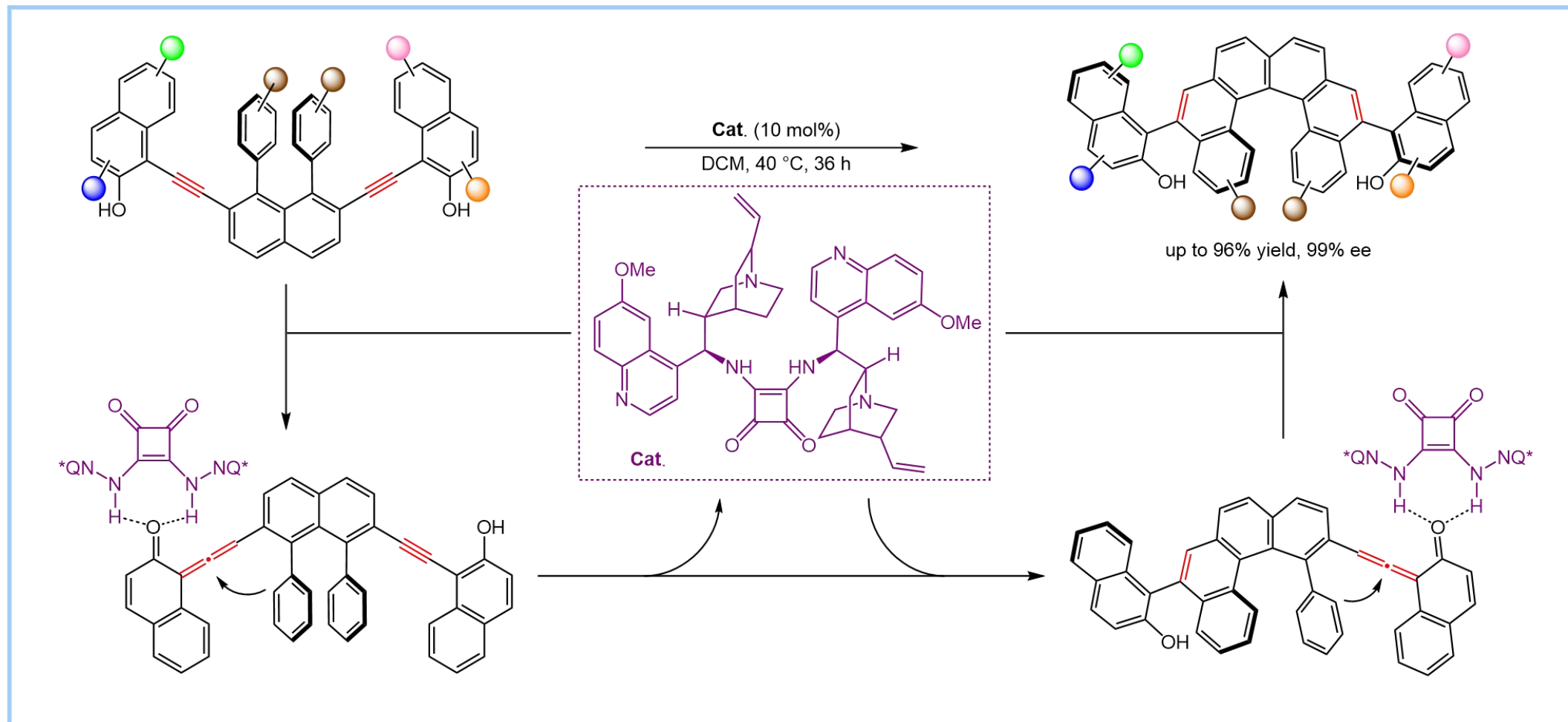
## Fisher indole synthesis (*CPA* catalysis)



Kotzner, L.; Webber, M. J.; Martinez, A.; Fusco, C. D.; List, B.\* *Angew. Chem. Int. Ed.* **2014**, *53*, 5202-5205

# Organocatalytic Synthesis of Helicenes

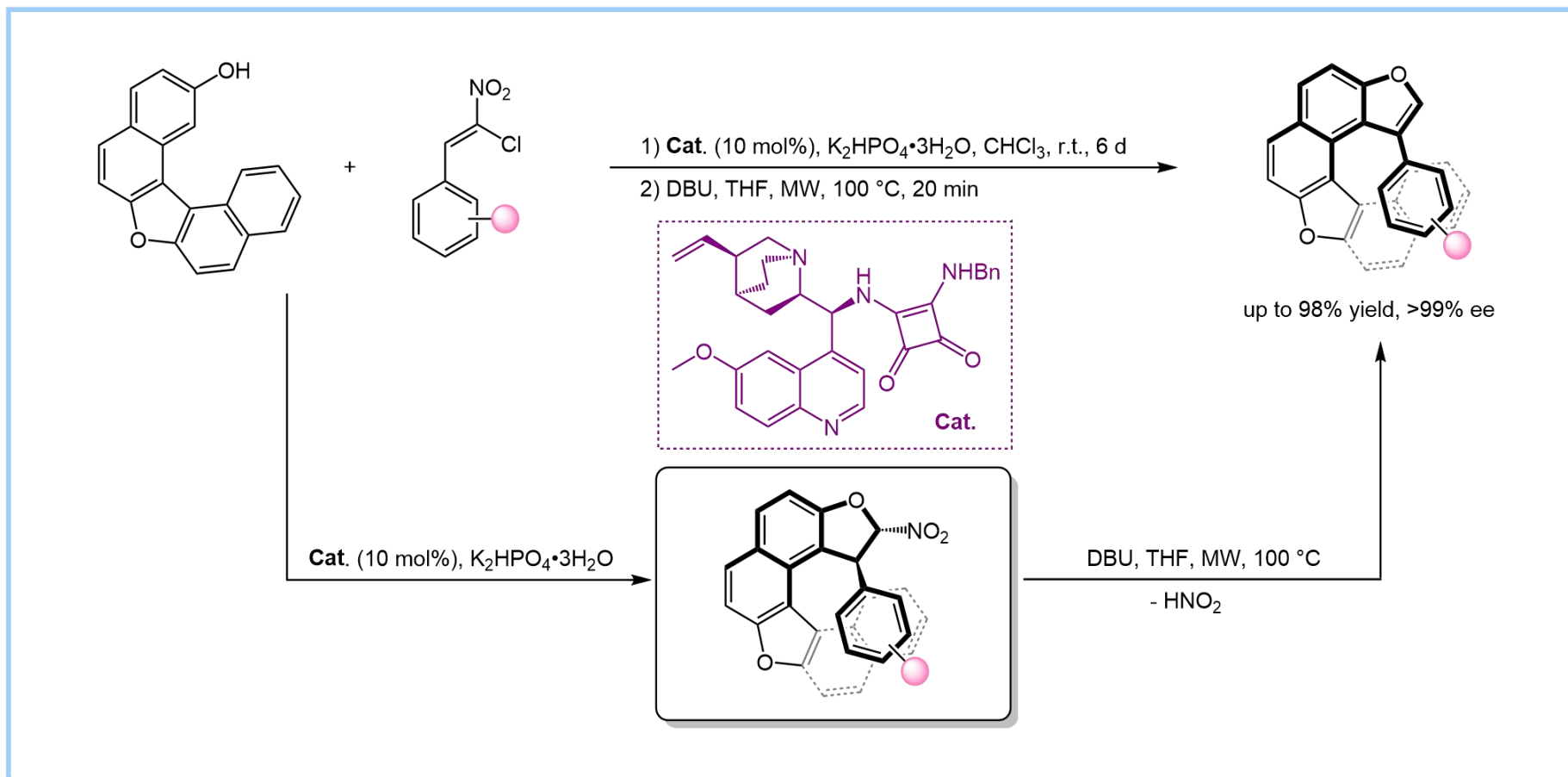
## [4+2] Cyclization (*Chiral squaramide catalysis*)



Jia, S.; Li, S.; Liu, Y.; Qin, W.; Yan, H.\* *Angew. Chem. Int. Ed.* **2019**, *58*, 18496-18501

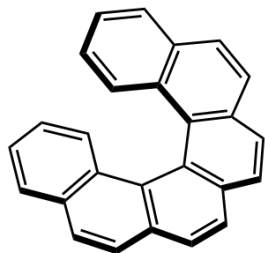
# Organocatalytic Synthesis of Helicenes

## [3+2] Cyclization (*Chiral squaramide catalysis*)

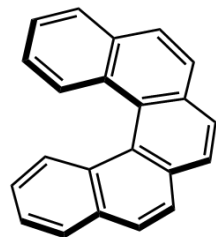


Liu, P.; Bao, X.; Naubron, J.-V.; Chentouf, S.; Humble, S.; Vanthuyne, N.; Bonne, D.\* *J. Am. Chem. Soc.* **2020**, *142*, 16199-16240

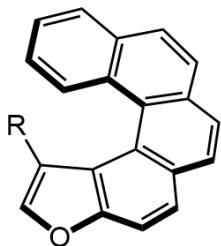
# Introduction



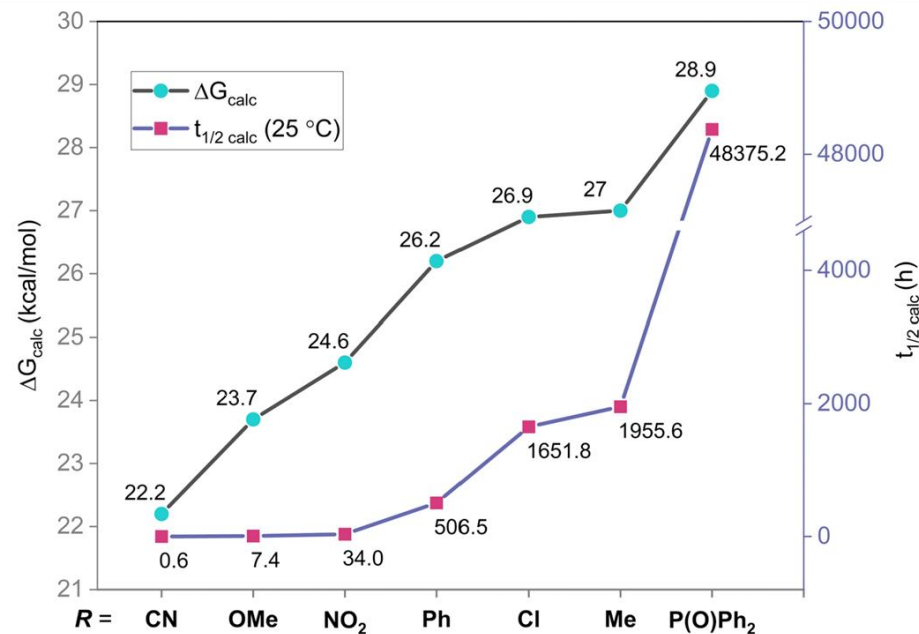
**carbo[6]helicene**  
 $\Delta G_{\text{exp}}^{\ddagger} = 35.9$  kcal/mol  
 $\Delta G_{\text{calc}}^{\ddagger} = 36.9$  kcal/mol  
configurationally **stable**



**carbo[5]helicene**  
 $\Delta G_{\text{exp}}^{\ddagger} = 24.8$  kcal/mol  
 $t_{1/2}^{25}$  calc = 47.5 h  
relatively **stable**

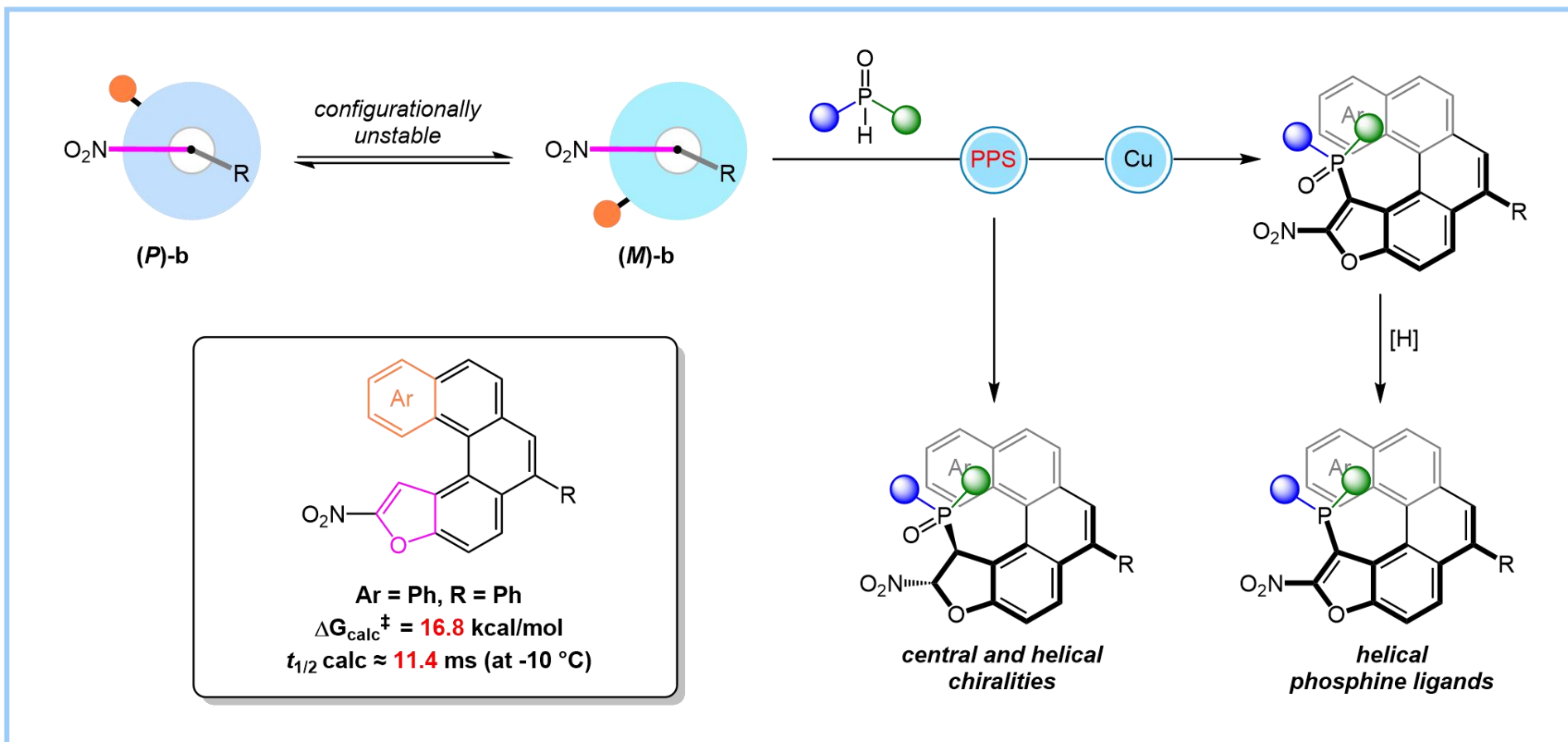


**oxa[5]helicene**  
 $\Delta G_{\text{calc}}^{\ddagger} = 15.7$  kcal/mol  
 $t_{1/2}^{25}$  calc = 36.4 ms (R = H)  
**unstable**



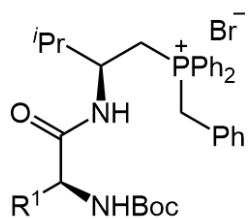
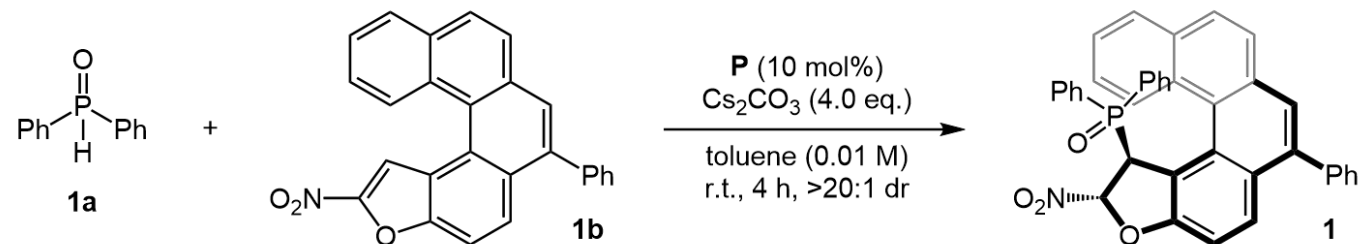
DFT and experiment guided relationship between conformational stability and structures

# Project Synopsis

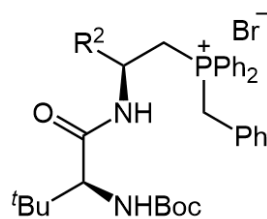




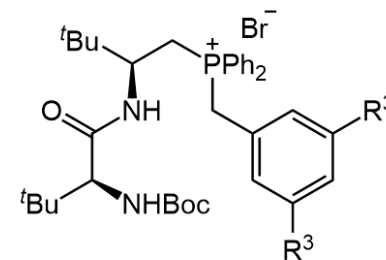
# Optimization of Reaction Conditions



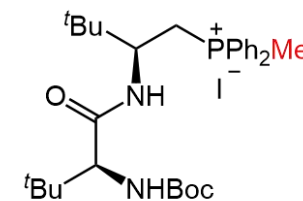
**P1:** R<sup>1</sup> = Me, 84% yield, 26% ee  
**P2:** R<sup>1</sup> = *i*Pr, 88% yield, 46% ee  
**P3:** R<sup>1</sup> = *t*Bu, 85% yield, 61% ee



**P4:** R<sup>2</sup> = Me, 87% yield, 44% ee  
**P5:** R<sup>2</sup> = *t*Bu, 87% yield, 80% ee

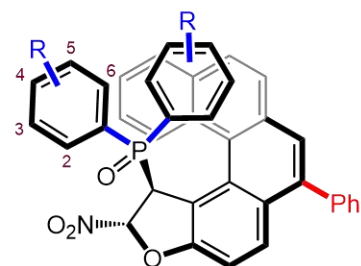
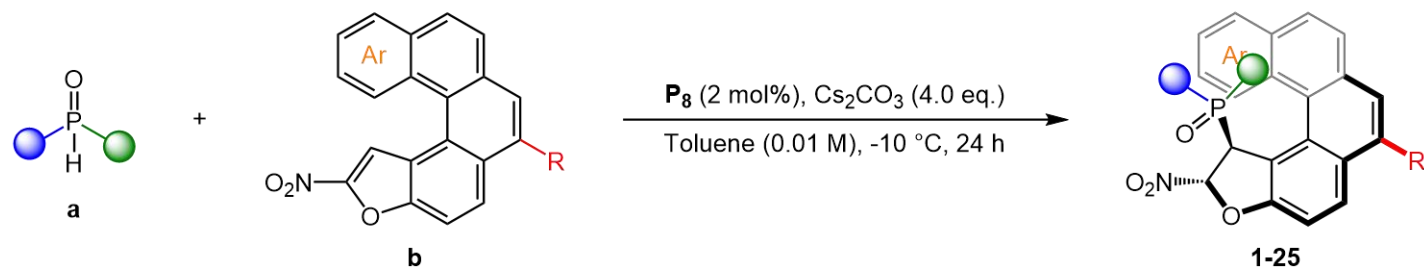


**P6:** R<sup>3</sup> = Me, 85% yield, 78% ee  
**P7:** R<sup>3</sup> = 3,4,5-OMe-C<sub>6</sub>H<sub>2</sub>, 87% yield, 81% ee



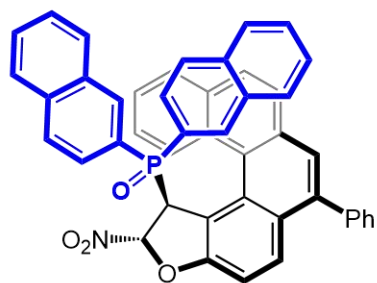
**P8**  
88% yield, 88% ee  
90% yield, 91% ee (0 °C, 10 h)  
92% yield, 92% ee (-10 °C, 24 h)  
92% yield, 92% ee (-10 °C, 24 h, 2 mol%)

# Substrate Scope

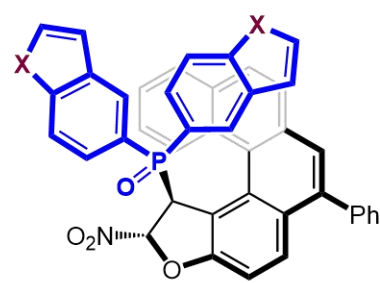


- 1, R = H, 92% yield, 92% ee
- 2, R = 3-Cl, 99% yield, 96% ee
- 3, R = 3-OMe, 98% yield, >99% ee
- 4, R = 4-F, 97% yield, 94% ee
- 5, R = 4-Cl, 99% yield, 92% ee
- 6, R = 4-OMe, 96% yield, 93% ee
- 7, R = 4-OEt, 96% yield, 96% ee

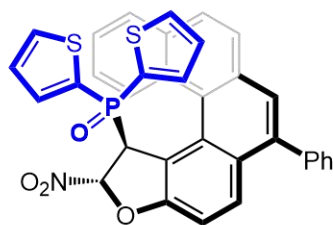
- 8, R = 4-OPh, 94% yield, 90% ee
- 9, R = 4-Me, 80% yield, 92% ee
- 10, R = 4-Et, 82% yield, 87% ee
- 11, R = 4-Ph, 97% yield, 97% ee
- 12, R = 3,5-Me, 98% yield, >99% ee
- 13, R = 3,5-OMe, 99% yield, >99% ee



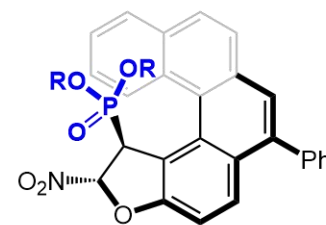
14, 99% yield, >99% ee



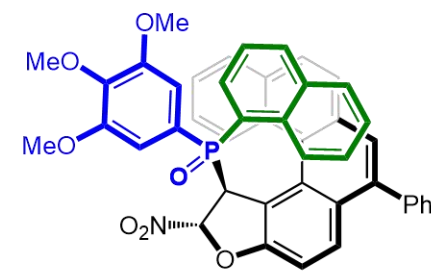
15, X = O, 98% yield, 96% ee  
16, X = S, 97% yield, 98% ee



17, 88% yield, 78% ee

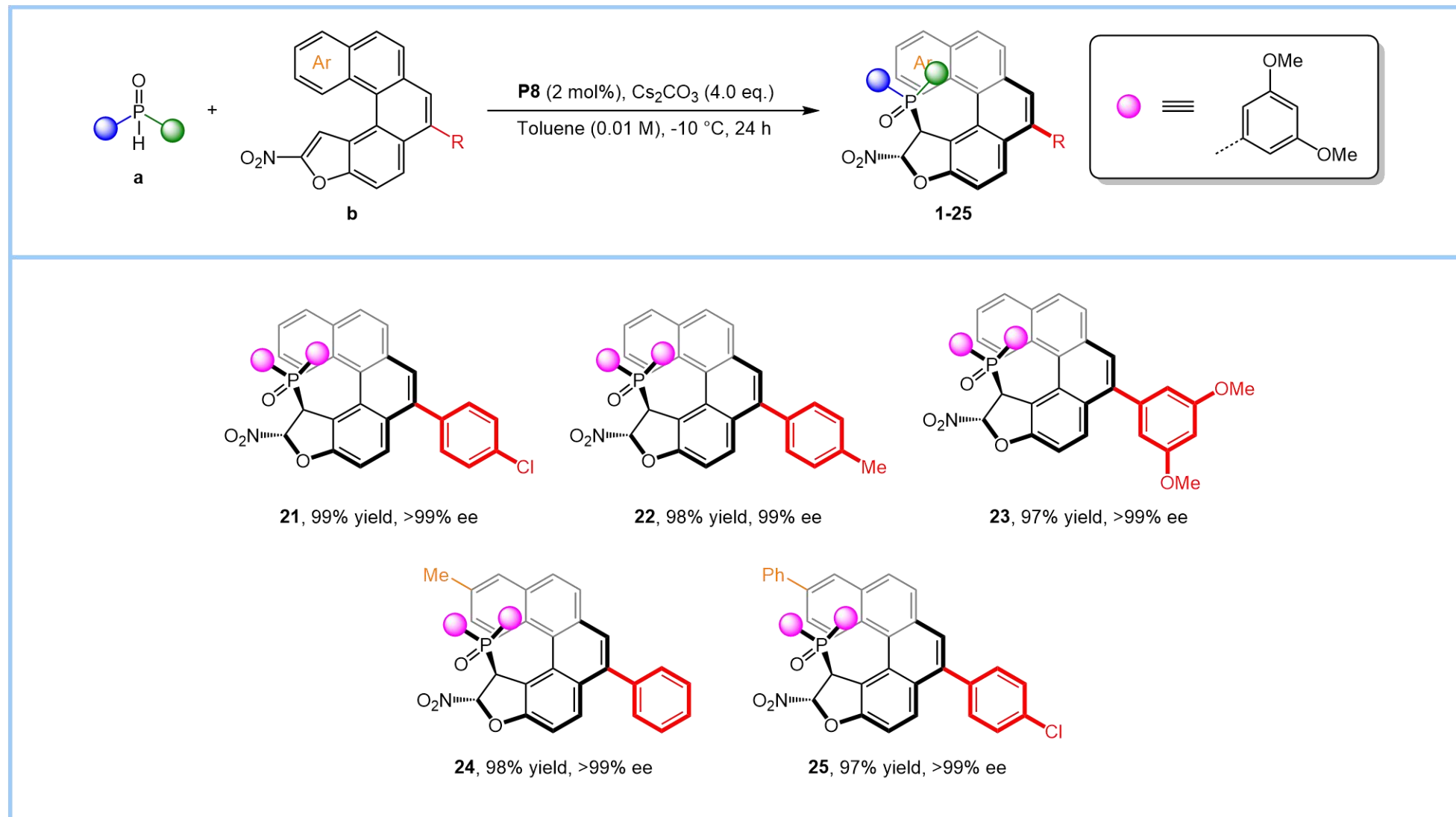


18, R = Me, 88% yield, 92% ee  
19, R = Et, 69% yield, 80% ee

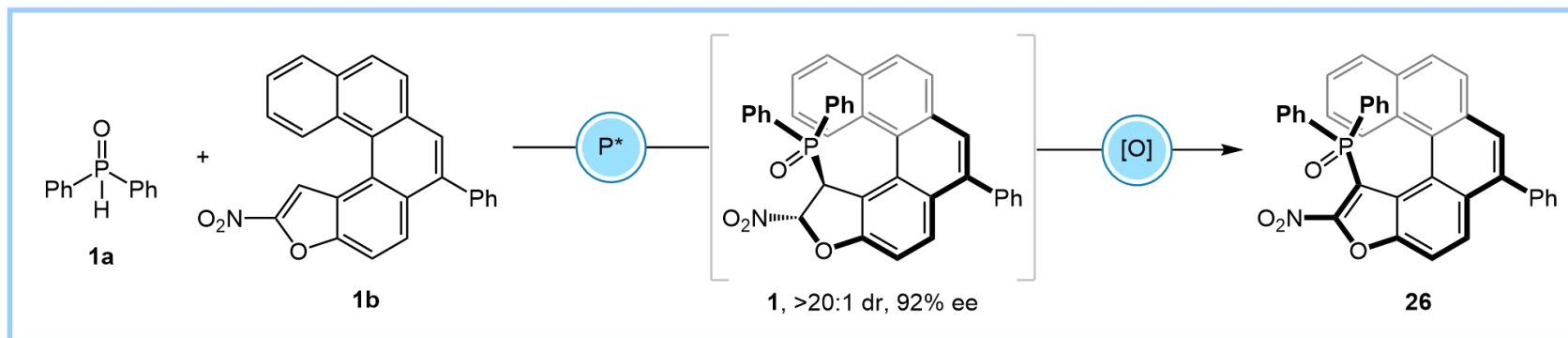


20  
89% yield, 5:1 dr, 97% ee (major)

# Substrate Scope



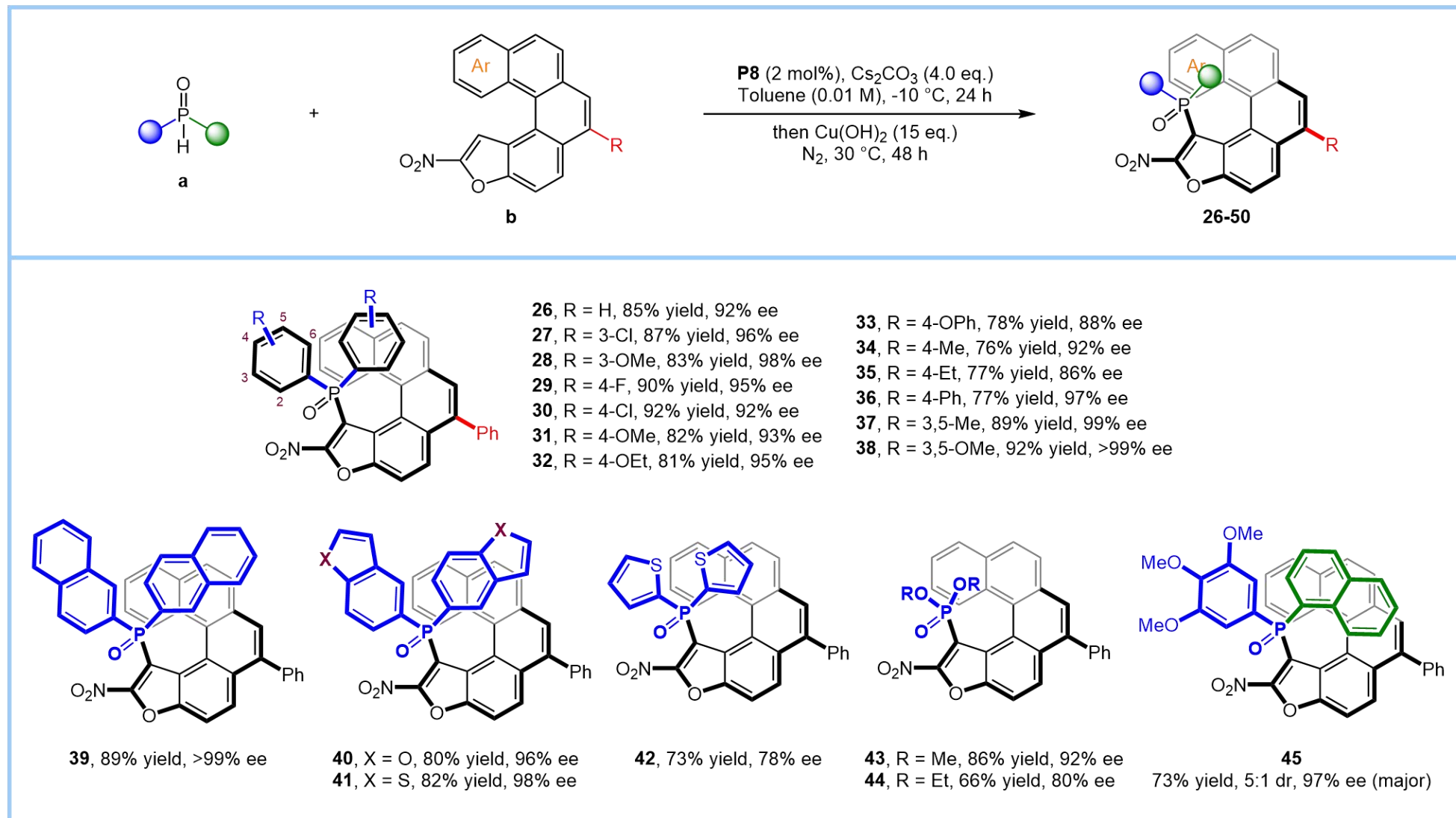
# Optimization of Reaction Conditions



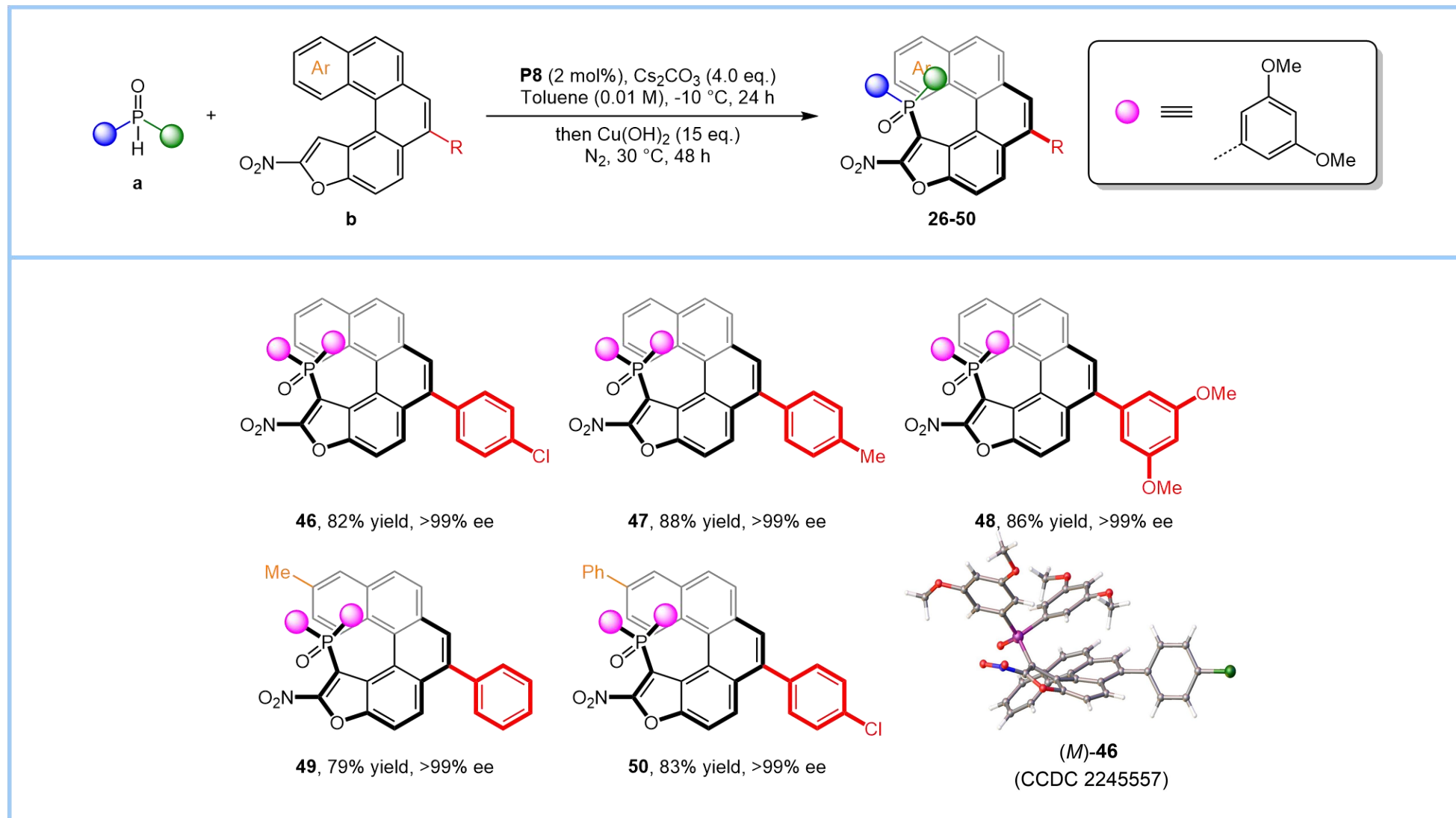
Entry <sup>[a]</sup>	[O]	Solvent	T (°C) / (h)	Yield (%)	Ee (%)
1	MnO <sub>2</sub>	CH <sub>2</sub> Cl <sub>2</sub>	0 (48)	<5	-
2	PhI(OAc) <sub>2</sub>	CH <sub>3</sub> Cl	0 (48)	<5	-
3	DDQ	CH <sub>2</sub> Cl <sub>2</sub>	40 (48)	<5	-
4 <sup>[b]</sup>	CAN	MeCN/H <sub>2</sub> O	r.t. (8)	<5	-
5	Ag <sub>2</sub> O	CH <sub>2</sub> Cl <sub>2</sub>	r.t. (48)	<5	-
6	Cu(NO <sub>3</sub> ) <sub>2</sub>	CH <sub>2</sub> Cl <sub>2</sub>	r.t. (72)	11	92
7 <sup>[c]</sup>	Cu(OH) <sub>2</sub>	Toluene	30 (48)	85	92

[a] Condition: **1a** (0.12 mmol), **1b** (0.1 mmol), **P8** (0.002 mmol), Cs<sub>2</sub>CO<sub>3</sub> (0.4 mmol) in 10 mL toluene at 10 °C for 24 h. After the reaction completed, the reaction mixture was filtered to remove Cs<sub>2</sub>CO<sub>3</sub>, and the residue was directly conducted under the following condition: oxidant (0.5 mmol) in 10 ml solvent. [b] MeCN/H<sub>2</sub>O (v/v = 4/1). [c] With Cu(OH)<sub>2</sub> (1.5 mmol) under N<sub>2</sub>.

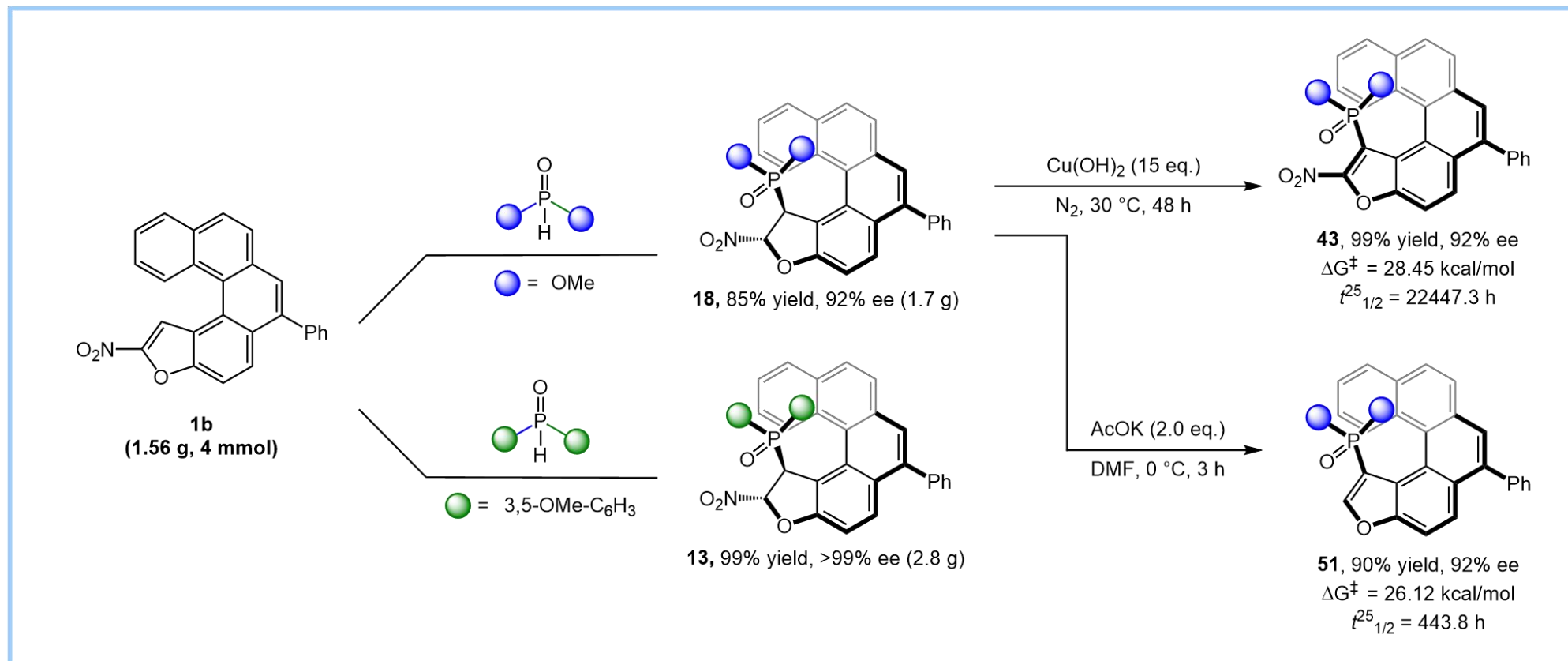
# Substrate Scope



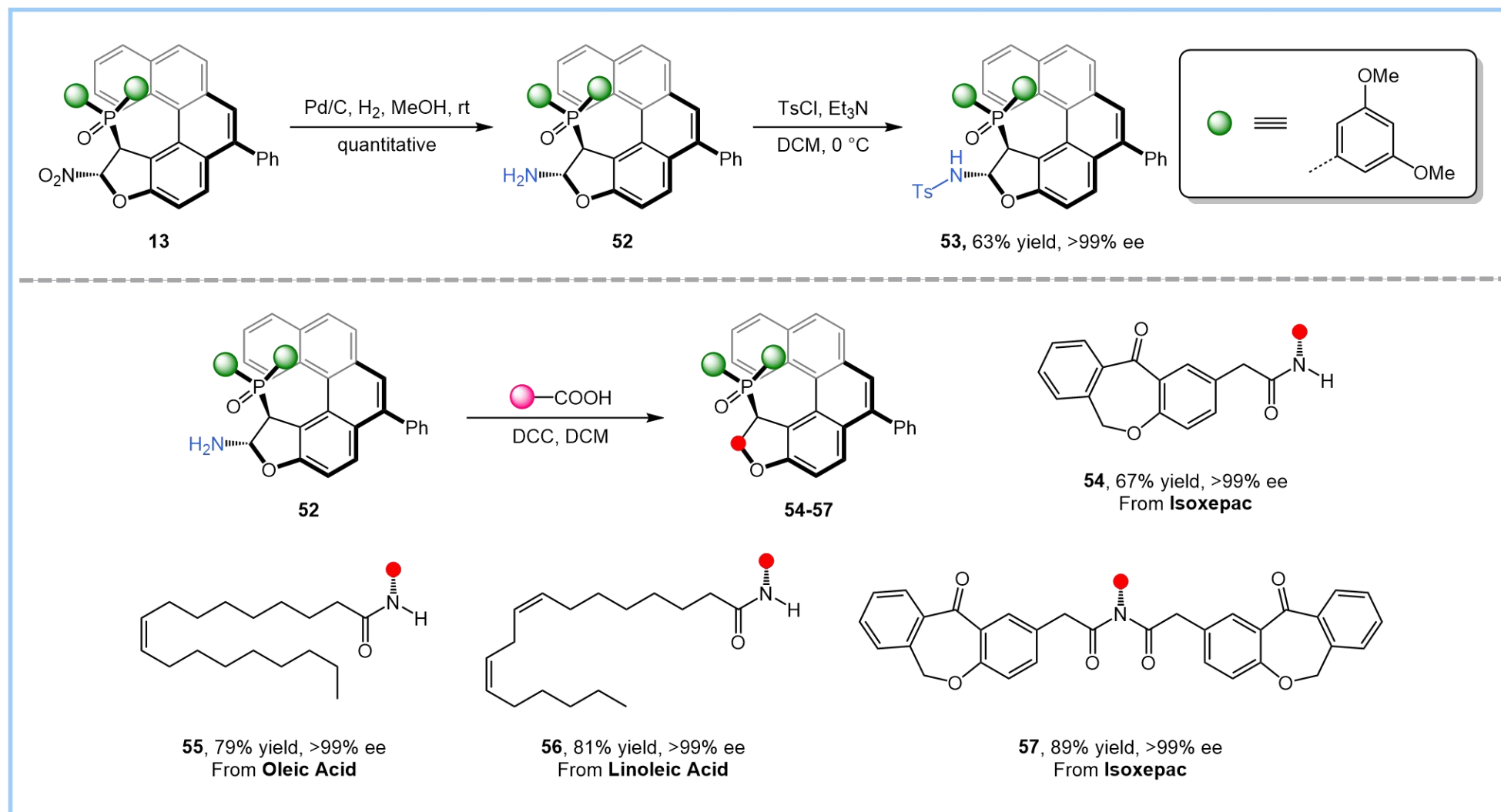
# Substrate Scope



# Gram Scale & Transformations of Products

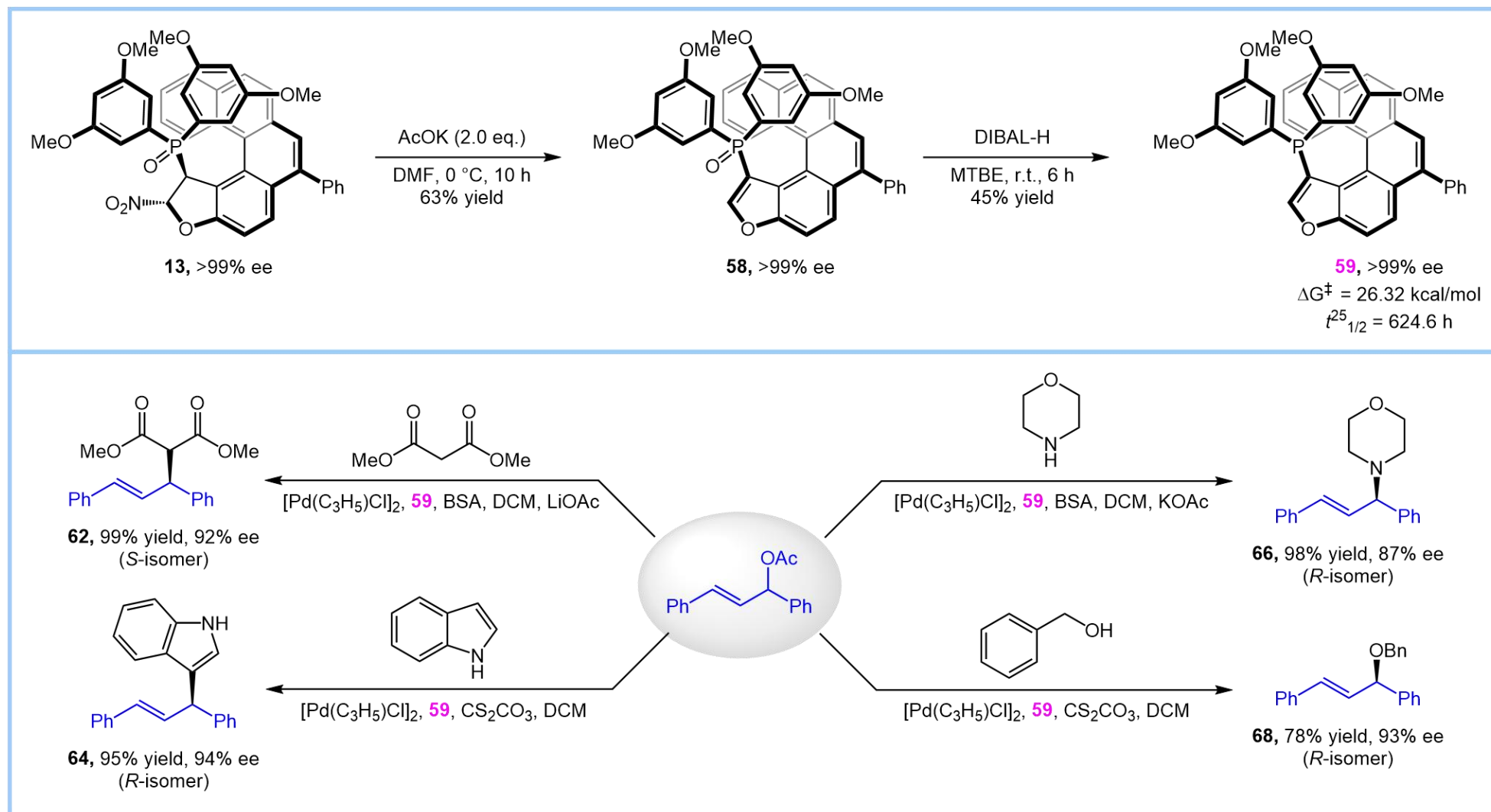


# Transformations of Products

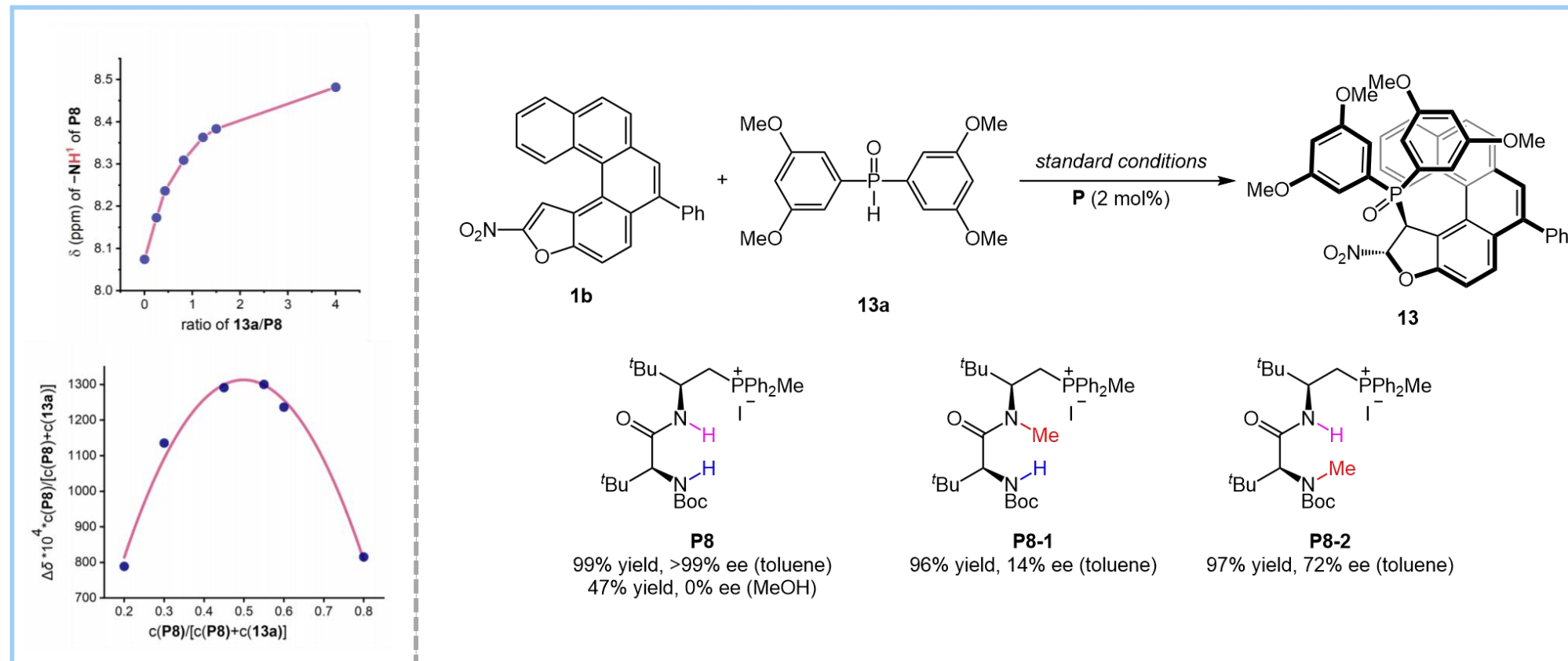




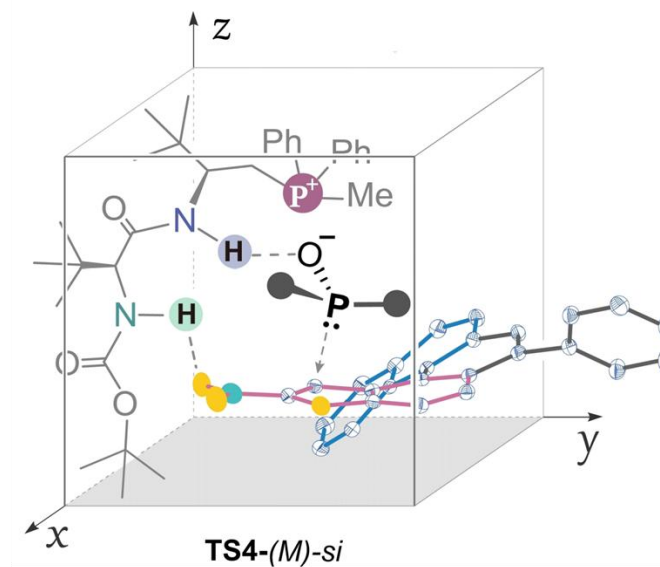
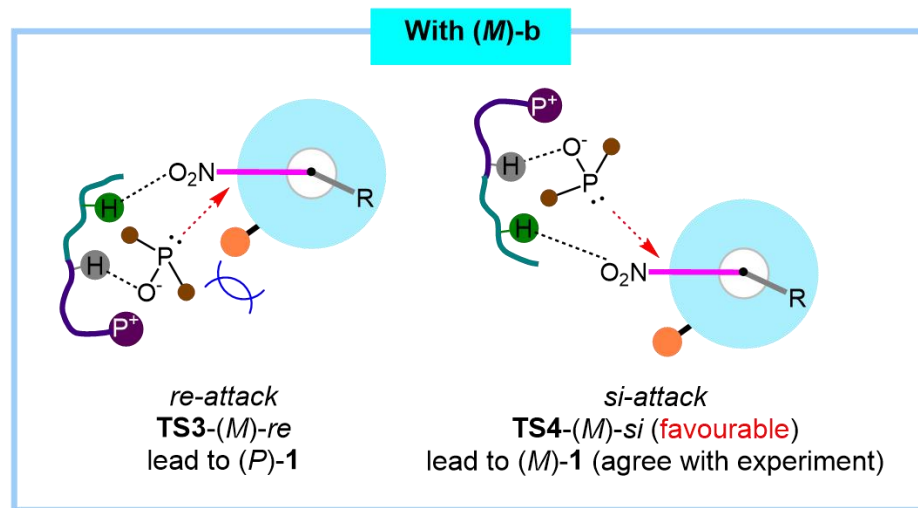
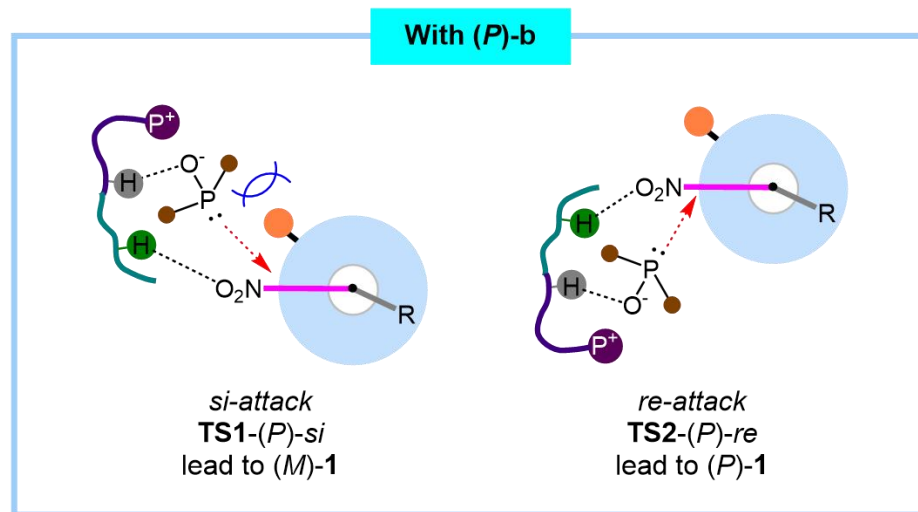
# Transformations of Products & Evaluation of 59



# Mechanism Study



# Mechanism Study



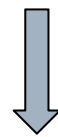


# The First Paragraph

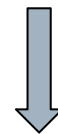
---

## 写作思路

螺烯类化合物具有重要的应用价值



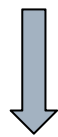
过去合成螺烯类化合物的方法



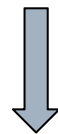
一步合成含磷的螺烯仍是一个挑战

## 写作思路

总结工作：一步构建含磷的螺烯类化合物



强调亮点：合成的含磷螺烯配体具有实用性



提出展望：进一步发展手性磷盐催化剂的应用

# Representative Examples

---

- Since their discovery by Witte and Meisenheimer in 1903, such **distinctive** topologies have greatly fascinated chemists due to their superiority in material sciences. (**distinctive**: 独特的)
- **Amongst them**, transition metal-catalyzed enantioselective synthesis of these scaffolds is widely considered the most general approach to date. (**Amongst them**: 在……之中)
- These limited asymmetric protocols and **disproportionate** development vastly impose restrictions on their versatility in asymmetric catalysis. (**disproportionate**: 不成比例的, 不均衡的)

# Acknowledgment

---

***Thanks  
for your attention !***